

Elephant Toothpaste

Elephant toothpaste is awesome! It is special because each tiny foam bubble is filled with oxygen. The yeast acts as a catalyst (a helper) to remove the oxygen from the hydrogen peroxide. Since it does this very fast, it creates lots and lots of bubbles. The experiment creates a reaction called an exothermic reaction, which means it not only creates foam, it also creates heat! The foam produced is just water, soap, and oxygen.

Materials:

- 1/2 a cup of Hydrogen peroxide
- A packet of dry yeast (one packet is approximately 1/4 oz)
- 3 Tablespoons warm water
- Food coloring
- A cylinder (at least 500 ml) or you can use a bottle
- A tablespoon of dish soap (any will do)
- Safety goggles

Instructions:

- 1. Pour 1/2 cup hydrogen peroxide solution, 1/4 cup dishwashing soap, and a few drops of food coloring into the bottle.
- 2. Swish the bottle around to mix the ingredients. Set the bottle in a sink or outdoors or some other place where you won't mind getting wet foam everywhere.
- 3. In a separate container, mix a packet of active yeast with a little warm water.
- 4. Give the yeast about five minutes to activate before proceeding to the next step.
- 5. When you are ready to do the demo, pour the yeast mixture into the bottle. The reaction occurs immediately upon the addition of the yeast.
- 6. It will be hot, be careful.

Modifications/Variations:

- 1. Use more than one food coloring
- 2. Use a tray to contain the mess
- 3. Use a stronger Hydrogen peroxide (20-volume hydrogen peroxide)

